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Frédéric Liéby-Müller, Christophe Allais, Thierry Constantieux, Jean Rodriguez. Metal-free Michael addition initiated multicomponent oxidative cyclodehydration route to polysubstituted pyridines from 1,3-dicarbonyls. Chemical Communications, 2008, pp.4207-4209. 10.1039/b805680c . hal-00677037

HAL Id: hal-00677037

<https://hal.science/hal-00677037>

Submitted on 7 Mar 2012

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Metal-free Michael addition initiated multicomponent oxidative cyclodehydration route to polysubstituted pyridines from 1,3-dicarbonyls†

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Received (in Cambridge, UK) 3rd April 2008, Accepted 19th May 2008

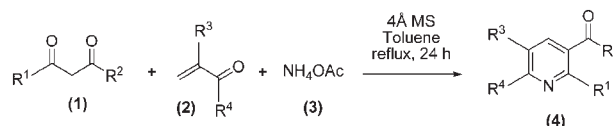
First published as an Advance Article on the web 10th July 2008

DOI: 10.1039/b805680c

A simple metal-free, step-economic and selective access to pyridines from readily available substrates is reported, involving a flexible 4 Å molecular sieves promoted Michael addition initiated domino three-component reaction between a 1,3-dicarbonyl, a Michael acceptor and a synthetic equivalent of ammonia.

Pyridines are one of the most important nitrogen heterocycles found in numerous natural and synthetic pharmaceutical agents.¹ These scaffolds are also of widespread interest in coordination and supramolecular chemistry, as well as for materials science.² The synthesis of these heterocycles has long been an area of intense interest resulting in the development of a wide range of synthetic methods.³ Among them, the direct condensation of carbonyl compounds with a source of ammonia is well documented,⁴ but still suffers from some limitations in the substrates,⁵ or involves an oxidative agent⁶ or an elimination step.⁷ Thereby, development of valuable synthetic pathways still remains an industrial as well as an academic challenge.⁸ In this context, the metal-catalysed [2+2+2] cycloisomerisation of alkynes with nitriles largely leads the way nowadays.⁹ However, despite recent spectacular advances,¹⁰ the low availability of some catalysts and substrates associated with the lack of regioselectivity¹¹ constitute major drawbacks.

In the course of our studies on the development of new domino¹² multicomponent reactions (MCRs)¹³ for creation of molecular complexity and diversity¹⁴ whilst combining economic aspects¹⁵ with environmental ones,¹⁶ we recently reported molecular sieves-promoted transformations of various 1,3-dicarbonyls¹⁷ for the stereoselective synthesis of a series of heterocycles.¹⁸ In this context, herein we wish to report on a simple metal-free, step-economic and selective access to pyridines from readily available substrates. Thus, we have now designed a flexible domino three-component reaction involving the direct condensation of 1,3-dicarbonyls **1** with Michael acceptors **2** and a synthetic equivalent of ammonia **3**, under heterogeneous catalysis by 4 Å molecular sieves (MS),



Scheme 1 MCR synthesis of polysubstituted pyridines **4**.

providing after *in situ* oxidation the corresponding pyridine derivatives **4** in a single operation (Scheme 1).¹⁹

Due to the nature of the three partners, this strategy may be viewed as a Michael addition initiated biomimetic approach previously formulated by Baldwin and Marazano²⁰ for natural 3-alkylpyridinium salts.

Preliminary experiments were conducted with easily available acyclic 1,3-dicarbonyls **1a–e** and Michael acceptors **2a–c**. Under optimised conditions, NH₄OAc proved to be the best source of ammonia²¹ and the corresponding pyridines **4a–j** were obtained by simply heating a toluene solution of the three partners in the presence of 4 Å MS,²² acting both as dehydrating agent and as heterogeneous catalyst as shown before.^{18a} The general applicability is clearly seen from the results reported in Table 1. Acrolein (**2a**) (entries 1, 4, 8, 10) and methacrolein (**2b**) (entries 2, 5, 7) may be used, as well as methyl vinyl ketone (**2c**) (entries 3, 6, 9). Similarly, diversity may be accessed through the use (Fig. 1) of either acetylacetone (**1a**) (entries 1–3), methyl acetoacetate (**1b**) (entries 4–6) or ethyl 4,4,4-trifluoroacetoacetate (**1c**) (entry 7). Interestingly enough, β-ketoamide **1d** led to the expected pyridines **4h** and **4i** (entries 8 and 9), making this transformation a direct and user-friendly one-pot access to nicotinamide derivatives. Finally, this multicomponent reaction appears as a promising new strategy for the direct metal-free synthesis of bi-aryl

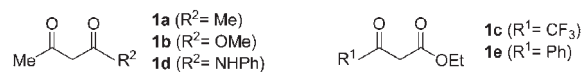


Fig. 1 Acyclic 1,3-dicarbonyl substrates **1** for the MCR.

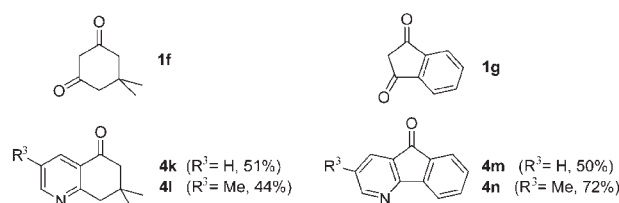


Fig. 2 Bi- and tricyclic pyridines from the MCR.

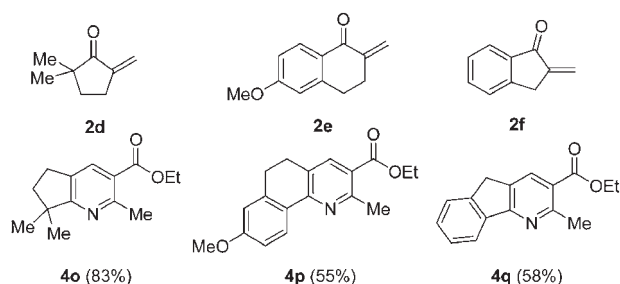
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† Electronic supplementary information (ESI) available: Complete experimental procedures and characterisations. See DOI: 10.1039/b805680c

Table 1 Pyridine synthesis from acyclic 1,3-dicarbonyls

Entry	Substrate 1	R ³	R ⁴	Product	Yield (%) ^a
1	1a	H	H	4a	52
2	1a	Me	H	4b	65
3	1a	H	Me	4c	62
4	1b	H	H	4d	56
5	1b	Me	H	4e	44
6	1b	H	Me	4f	65
7	1c	Me	H	4g	70
8	1d	H	H	4h	61
9	1d	H	Me	4i	42
10	1e	H	H	4j	65

^a Isolated yield after flash chromatography.**Fig. 3** Pyridines from sensitive Michael acceptors.

compounds from substrates such as **1e** (entry 10), opening the way to a flexible design of atropoisomers of bi-aryl ligands.²³

To further demonstrate the versatility of the method, we then examined the use of cyclic 1,3-dicarbonyls such as dimedone (**1f**) or indane-1,3-dione (**1g**) in the sequence, and some representative bi- and tricyclic pyridines are shown in Fig. 2. In all cases, products are obtained with a total regioselectivity. Of particular interest is the one-pot synthesis of 4-azafluorenones **4m** and **4n**, which are common skeletons in natural products and molecules of pharmacological interest,²⁴ and generally accessed *via* multistep sequences.²⁵

The neutral heterogeneous reaction conditions are also suitable with sensitive Michael acceptors such as α -*exo*-methylene ketones **2d–f**,²⁶ leading to bi- and tricyclic pyridines **4o–q** in acceptable yields (Fig. 3).

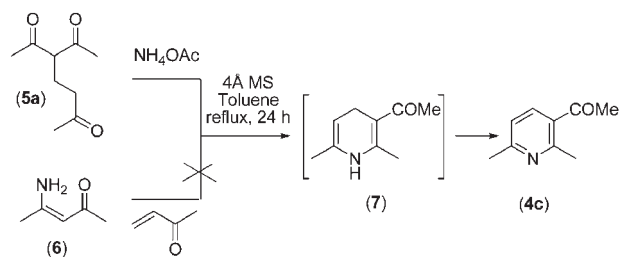
From a mechanistic point of view, two multistep sequences have been preliminarily explored. Both evolve through a 1,4-dihydropyridine intermediate **7** which suffers an *in situ* oxidative aromatisation to the corresponding pyridine.²⁷ We initially postulated that the first step of the sequence may be the molecular sieves promoted Michael addition between substrates **1** and acceptors **2**. The corresponding adduct **5**

may then react with ammonium acetate (**3**) leading to the dihydropyridine **7** *via* an intramolecular dehydrative cyclisation sequence. As a validation of this first hypothesis, pyridine **4c** was isolated by mixing the Michael adduct **5a**²⁸ with **3** under standard conditions (Scheme 2). Alternatively, a more conventional mechanistic pathway could involve the preliminary formation of an enamino ketone intermediate **6**, which may lead to the final product *via* a Hantzsch-type reaction.²⁹ Interestingly enough, when **6**, independently prepared from NH₄OAc and acetylacetone (**1a**), was reacted with methyl vinyl ketone (**2c**), pyridine **4c** was not formed and starting materials were recovered even after 24 hours (Scheme 2). These preliminary results support our original mechanistic proposal involving a 4 Å MS initiated Michael addition³⁰ as the first step of the sequence.³¹

In conclusion, we have developed a regioselective, user-friendly and mechanistically original three-component reaction for the one-pot synthesis of polysubstituted pyridines from readily accessible substrates. The biomimetic like sequence does not require any harmful reagents or metal-based catalysts, and allows construction of highly functionalised heterocycles of both biological and synthetic interest. This pyridine approach should be a good and complementary substrate directed synthetic alternative to other well known methods.

Notes and references

- (a) H. J. Roth and A. Kleemann, *Drug Synthesis*, in *Pharmaceutical Chemistry*, John Wiley and Sons, New York, 1988, vol. 1; (b) J. P. Michael, *Nat. Prod. Rep.*, 2005, **22**, 627–646.
- M. Balasubramanian and J. G. Keay, *Comprehensive Heterocyclic Chemistry II*, ed. A. R. Katritzky, C. W. Rees and E. F. V. Scriven, Pergamon, Oxford, 1996, vol. 5, pp. 245–300.
- D. Spitzner, in *Science of Synthesis*, ed. D. Black, StC., Georg Thieme Verlag, Stuttgart, 2005, pp. 11–284.
- (a) A. E. Chichibabin and O. A. Zeide, *J. Russ. Phys. Chem.*, 1906, **37**, 1229; (b) F. Bohlmann and D. Rahtz, *Chem. Ber.*, 1957, **90**, 2265–2272; (c) H. Nozaki, S. Fujita and T. Mori, *Bull. Chem. Soc. Jpn.*, 1969, **42**, 1163.
- (a) M. C. Bagley, C. Glover and E. A. Merritt, *Synlett*, 2007, 2459–2482, and references cited therein; (b) A.-L. Blayo, S. Le Meur, D. Grée and R. Grée, *Adv. Synth. Catal.*, 2008, **350**, 471–476.
- See for example: (a) J. S. Yadav, B. V. S. Reddy, A. K. Basak, G. Baishya and A. V. Narsaiah, *Synthesis*, 2006, 451–454; (b) T. R. K. Reddy, R. Mutter, W. Heal, K. Guo, V. J. Gillet, S. Pratt and B. Chen, *J. Med. Chem.*, 2006, **49**, 607–615.

**Scheme 2** Mechanistic investigations.

- 7 (a) F. Kröhnke, *Synthesis*, 1976, 1–24; (b) G. J. Reddy, D. Latha, C. Thirupathiah and K. S. Rao, *Tetrahedron Lett.*, 2005, **46**, 301–302; (c) S. Kantevari, M. V. Chary and S. V. N. Vuppapalapati, *Tetrahedron*, 2007, **63**, 13024–13031.
- 8 (a) F. Mongin and G. Queguiner, *Tetrahedron*, 2001, **57**, 4059–4090; (b) G. D. Henry, *Tetrahedron*, 2004, **60**, 6043–6061; (c) A. R. Katritzky, *Chem. Rev.*, 2004, **104**, 2125–2126.
- 9 (a) J. A. Varela and C. Saá, *Chem. Rev.*, 2003, **103**, 3787–3802; (b) I. Nakamura and Y. Yamamoto, *Chem. Rev.*, 2004, **104**, 2127–2198.
- 10 For recent examples with various metals, see: (a) Y. Zhou, J. A. Porco, Jr and J. K. Snyder, *Org. Lett.*, 2007, **9**, 393–396; (b) H. T. Chang, M. Jegannathan and C.-H. Cheng, *Org. Lett.*, 2007, **9**, 505–508; (c) K. Kase, A. Goswami, K. Ohtaki, E. Tanabe, N. Saino and S. Okamoto, *Org. Lett.*, 2007, **9**, 931–934; (d) A. Wada, K. Nogushi, M. Hirano and K. Tanaka, *Org. Lett.*, 2007, **9**, 1295–1298.
- 11 (a) C. Brändli and T. R. Ward, *J. Comb. Chem.*, 2000, **2**, 42–47; (b) K. Parthasarathy, M. Jegannathan and C.-H. Cheng, *Org. Lett.*, 2008, **10**, 325–328.
- 12 (a) *Domino Reactions in Organic Synthesis*, ed. L. F. Tietze, G. Brasche and K. M. Gericke, Wiley-VCH, Weinheim, 2006; (b) A. Padwa and S. K. Bur, *Tetrahedron*, 2007, **63**, 5341–5378.
- 13 (a) *Multicomponent Reactions*, ed. J. Zhu and H. Bienaymé, Wiley-VCH, Weinheim, 2005; (b) A. Dömling, *Chem. Rev.*, 2006, **106**, 17–89.
- 14 (a) T. E. Nielsen and S. L. Schreiber, *Angew. Chem., Int. Ed.*, 2008, **47**, 48–56; (b) For a recent example of structurally-diversifying domino reaction sequences yielding highly substituted pyridines, see: H. Waldmann, M. Kühn, W. Liu and K. Kumar, *Chem. Commun.*, 2008, 1211–1213.
- 15 (a) Step-economy: P. A. Wender, G. G. Gamber, R. D. Hubbard, S. M. Pham and L. Zhang, *J. Am. Chem. Soc.*, 2005, **127**, 2836–2837; (b) Atom economy: B. M. Trost, *Acc. Chem. Res.*, 2002, **35**, 695–705.
- 16 Special issue in green chemistry: *Chem. Rev.*, 2007, **107**, 2167–2820.
- 17 (a) C. Simon, T. Constantieux and J. Rodriguez, *Eur. J. Org. Chem.*, 2004, 4957–4980; (b) F. Liéby-Muller, C. Simon, T. Constantieux and J. Rodriguez, *QSAR Comb. Sci.*, 2006, **25**, 432–438.
- 18 (a) C. Simon, J. F. Peyronel and J. Rodriguez, *Org. Lett.*, 2001, **3**, 2145–2148; (b) F. Liéby-Muller, T. Constantieux and J. Rodriguez, *J. Am. Chem. Soc.*, 2005, **127**, 17176–17177; (c) F. Liéby-Muller, T. Constantieux and J. Rodriguez, *Synlett*, 2007, 1323–1325.
- 19 For gas phase utilisation of zeolites as heterogeneous catalyst, see: W. Hoelderich and N. Goetz, *US Pat.*, 4960894, 1990.
- 20 (a) J. E. Baldwin and R. C. Whitehead, *Tetrahedron Lett.*, 1992, **33**, 2059–2062; (b) A. Kaiser, X. Billot, A. Gateau-Olesker, C. Marazano and B. C. Das, *J. Am. Chem. Soc.*, 1998, **120**, 8026–8034; (c) J.-C. Wypych, T. M. Nguyen, M. Bénéchie and C. Marazano, *J. Org. Chem.*, 2008, **73**, 1169–1172.
- 21 (a) Sulfamic acid^{21b} led also to the formation of pyridines, but with lower yields. Hydroxylamine and ammonium chloride were tested without any success; (b) C. Defieber, M. A. Ariger, P. Moriel and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2007, **46**, 3139–3143.
- 22 Crude products were easily obtained with acceptable chemical purity by simple filtration through a short pad of Celite. It is noteworthy that no reaction occurs in the absence of molecular sieves.
- 23 G. Bringmann, A. J. Price Mortimer, P. A. Keller, M. J. Gresser, J. Garner and M. Breuning, *Angew. Chem., Int. Ed.*, 2005, **44**, 5384–5427.
- 24 K. J. Stauffer, P. D. Williams, H. G. Selnick, P. G. Nantermet, C. L. Newton, C. F. Homnick, M. M. Zrada, S. D. Lewis, B. J. Lucas, J. A. Krueger, B. L. Pietrak, E. A. Lyle, R. Singh, C. Miller-Stein, R. B. White, B. Wong, A. A. Wallace, G. R. Sitko, J. J. Cook, M. A. Holahan, M. Stranieri-Michener, Y. M. Leonard, J. J. Lynch, D. R. McMasters and Y. Yan, *J. Med. Chem.*, 2005, **48**, 2282–2293.
- 25 See for example: (a) A.-S. Rebstock, F. Mongin, F. Trecourt and G. Queguiner, *Tetrahedron*, 2003, **59**, 4973–4977; (b) Multicomponent approach: S. Tu, B. Jiang, H. Jiang, Y. Zhang, R. Jia, J. Zhang, Q. Shao, C. Li, D. Zhou and L. Cao, *Tetrahedron*, 2007, **63**, 5406–5414.
- 26 (a) J.-L. Gras, *Tetrahedron Lett.*, 1978, **24**, 2111–2114; (b) J.-L. Gras, *Org. Synth.*, 1981, **60**, 88.
- 27 The poor stability of dihydropyridines in acetic conditions is well known and has been recently exemplified as a side reaction in the use of Hantzsch ester as a reducing agent: G. Barbe and A. B. Charette, *J. Am. Chem. Soc.*, 2008, **130**, 18–19.
- 28 **5a** was synthesised from acetylacetone (**1a**) and methyl vinyl ketone (**2c**) with a catalytic amount of base-supported P-BEMP: D. Bensa, T. Constantieux and J. Rodriguez, *Synthesis*, 2004, 923–927.
- 29 A. Hantzsch, *Justus Liebigs Ann. Chem.*, 1882, 215–236.
- 30 After optimisation of the reaction conditions, we demonstrated that 2 equiv. of NH₄OAc (**3**) were necessary. However, it is noteworthy that when only 1 equiv. of **3** was used, the expected pyridine was obtained in admixture with a significant amount of **5**, which may confirm the preliminary formation of such Michael adduct intermediates in the first step of the sequence.
- 31 The catalytic properties of 3 Å MS activated with several neutral Lewis bases were also recently reported for the promotion of Michael addition of 1,3-dicarbonyl compounds: R. Villano and A. Scettri, *Synthesis*, 2005, 757–760.